Time = 1200.0 seconds.

Compartment	Upper Temp. (K)	Lower Temp. (K)		Upper Vol. (m^3)		Upper Absorb (m^-1)	Lower Absorb (m^-1)		Ambient Target (W/m^2)	Floor Target (W/m^2)
1	875.5	439.1	0.7806	15. (68%)	0.000E+00	0.000E+00	-4.19	3.343E+04	3.998E+04
Fires										
Compartment	Fire	Plume Flow (kg/s)	Pyrol Rate (kg/s)	Fire Size (W)	Flame Heigh (m)		_		Convec.	Radiat. (W)
1	Main	1.57 1.57		1.047E+06		=	E-10 1.047	E+06 0.000E+0		5 3.140E+05

Flow Through Vents (kg/s)

To	Through	Upper Layer		Lower Layer		Mixing	Mixing
Compartment	Vent	Inflow	Outflow	Inflow	Outflow	To Upper	To Lower
1	H Outside #1 H Outside #2		6.662E-03	6.890E-03			
	H Outside #3		1.40	1.34			0.170
Outside	H Comp 1 #1	C CC2E 02			6.890E-03		
	H Comp 1 #2 H Comp 1 #3	6.662E-03 1.40			1.34		

The mass balance is just the sum of the flow in minus the flow out. For any compartment, this is just

("Upper Layer Inflow" + "Lower Layer Inflow" + "Pyrol Rate") – ((Upper Layer Outflow + Lower Layer Outflow) with the inflow and outflow summed for each vent. Note that the mixing flows are ignored since they are just exchanges between layers in the same compartment. For the above example, it's just

$$(6.890E-03 + 1.34 + 5.409E-02) - (6.662E-03 + 1.40) = -0.005682 \text{ kg/s}$$

or less than 1/2 percent of the magnitude of the inflow. In theory, at a perfect steady-state solution, this number should be zero. In practice, it never is, but it should be small compared to the flow.